

Journal of Chemical, Biological and Physical Sciences



An International Peer Review E-3 Journal of Sciences

Available online at www.jcbps.org

Section A: Chemical Sciences

CODEN (USA): JCBPAT

Research Article

An efficient one-pot multi Component Synthesis of pyrimidine and their Antimicrobial activity

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Received: 28 November 2016; **Revised:** 16 December 2016; **Accepted:** 23 December 2016

Abstract: One pot three component synthesis of 4-amino/mercapto-6-(3-nitrophenyl)-2-sulfanylpyrimidine-5-carbonitrile (IVa-b) in moderate to high yield is achieved via a one pot synthesis reaction of malononitrile, substituted aromatic aldehyde & Guanidine/thiourea in ethanol. Which is further treated with formic acid via Cyclization gives 7-Substituted-5-(3-nitrophenyl) pyrimido [4,5-*d*]pyrimidin-4(3*H*)-one (V a-b) in condensation process. Which was further treated with Bromoethoxyphthalimide to give 2-(2-{[4-(3-nitrophenyl)-5-oxo-5,6-dihydropyrimido[4,5-*d*]pyrimidin-2-yl]sulfanyl/ amino} ethoxy)-1*H*-isoindole-1,3(2*H*)-dione (VIIIa-b). All the synthesized compounds give were supported by spectral and analytical analysis. Also the synthesized compounds were tested for their antimicrobial activity.

Keywords: Malononitrile, Thiourea, Guanidine, Pyrimidine, Tetrazolopyrimidine, s-Triazolopyrimidine.

INTRODUCTION

Heterocyclic chemistry evolved Medicinal chemistry is a scientific discipline at the intersection of chemistry and pharmacy involved with designing, synthesizing and developing pharmaceutical drugs. It also includes the study of existing drugs, their biological properties, and their quantitative structure

activity relationships (QSAR). New drugs are necessitated, to cure new diseases, to find harmless drugs (no side effects) and to cure diseases whose drugs have become ineffective due to resistant strains of micro-organisms. Besides these causes, new drug discovery and research in medicinal chemistry is required to identify pharmacophore present in effective drugs (as a little modification in the structure of a drug may cause abrupt change in its pharmacology). Research is necessary to replace the broad-spectrum medicines by narrow spectrum drugs, which can be achieved by designing the drug in the light of pharmacogenomics. Also, the global population of different physical and climatic conditions may not tolerate a single drug. Thus there is a growing need of research in the field of synthetic chemistry.

Pyrimidine ring as Tetrazolo pyrimidine play an important role as potent analgesic, anti-inflammatory, antipyretic, antimicrobial, anticonvulsant, fungicidal, antiplatelet activities, and central nervous system (CNS) affecting activities. In addition, pyrimidine nucleus can be found in a broad variety of antibacterial, anticancer and anti-tumor agents as well as in agrochemicals and veterinary products, Pyrimidine derivatives serve both as biomimetic and reactive pharmacophores due to their diverse medicinal properties such as antitumor, anticancer (lungs, breast and CNS cancer), immunodilator, antiviral etc. Pyrimidine derivatives have activities like tyrosine kinase inhibitors, COX-2 inhibitor, calcium channel blockers plus antihypertensive and also activity against Y181C HIV-1 mutant strain, etc. Diverse biological activities like: anticonvulsant, diuretic, fungicidal and trypanocidal. Pyrimidine derivatives possess several interesting biological activities such as antimicrobial, antitumour and antifungal activities¹⁻⁵.

RESULT & DISCUSSION

Our continuous research directed to synthesise biologically active compounds, we report here a new synthesis for 2-amino-, 2-mercapto- and 2-benzylmercaptopyrimidine derivatives *via* the reaction of aromatic aldehydes, malononitrile and each of Thiourea, Guanidine and S-benzylisothiurea respectively. The reaction products were utilised for the synthesis of other new azolopyrimidines bearing latent functional substituents. Thus, when a ternary mixture of the appropriate aromatic aldehyde, malononitrile and thiourea or guanidine was refluxed in ethanol in the presence of sodium acetate, the 2-mercapto and 2,4- diamino-5-cyano-6-phenyl pyrimidine (IVa-b) were obtained. Structure (IVa-b) was inferred from the following facts: The ¹H NMR spectrum of IVa showed signals at 3.32 (s, 1H, SH), 4.14 (s, 2H, NH₂), 7.61-7.96 (m, 4H, Ar-H) and IVb showed signals at 4.48 (s, 4H, NH₂) 7.63-7.96 (m, 4H, Ar-H) led to the disappearance of the first and the last signals. The IR spectra of IVa-b displayed characteristic bands for NH₂, CN and SH. **Compound VIa** showed the ¹H NMR signals at 2.42 (s, 2H, CH₂), 4.14 (s, 2H, NH₂), 7.61-7.96 (m, 9H, Ar-H). The IR spectra of VIa displayed characteristic bands for NH₂, CN and CH₂. **Compound Va** showed the ¹H NMR signals at 1.56 (s, 1H, CH) 3.32 (s, 1H, SH) 8.56 (s, 1H, NH), 7.63-7.96 (m, 4H, Ar-H), The IR spectra of Va displayed characteristic bands for NH, CN, CO, CH and SH. Whereas **Compound Vb** showed the ¹H NMR signals at 1.48 (s, 1H, CH) 4.56 (s, 2H, NH₂) 8.41 (s, 1H, NH), 7.26-7.91 (m, 4H, Ar-H). The IR spectra of Vb displayed characteristic bands for NH₂, NH, CN, SH, CO and CH.

All the synthesized compounds are tested for antibacterial and anti fungal activity. In these compounds **IVb, Va & Vb** shows good activity against bacterial and **Va & Vb** gave good activity against fungal and rest of compounds show moderate activity.

MATERIAL & METHOD

All the chemicals and solvents (analytical grade) were purchased from commercial sources and used without further purification. All melting points were determined in open capillary tube and are uncorrected. TLC aluminum sheets were used for thin-layer chromatography (TLC) and spots were visualized under UV light. Melting points were taken in open capillary tubes and therefore uncorrected. Purity of the compounds was checked on silica gel G TLC plates of 2 mm thickness using n-hexane and ethyl acetate as solvent system. The visualization of spots was carried out in an UV/Iodine chamber. The IR spectra of the compounds were recorded in the 4000-450 cm^{-1} ranges using KBr discs on FTIR Perkin Elmer spectrometers and ^1H NMR were recorded on a Bruker DRX-300 MHz spectrometer (DMSO) using TMS as an internal standard. The mass spectra were recorded on a Jeol SX-102 (FAB) mass spectrometer. Structure of all the synthesized compounds was assigned on basis of their analytical Data and spectral data.

EXPERIMENTAL SECTION

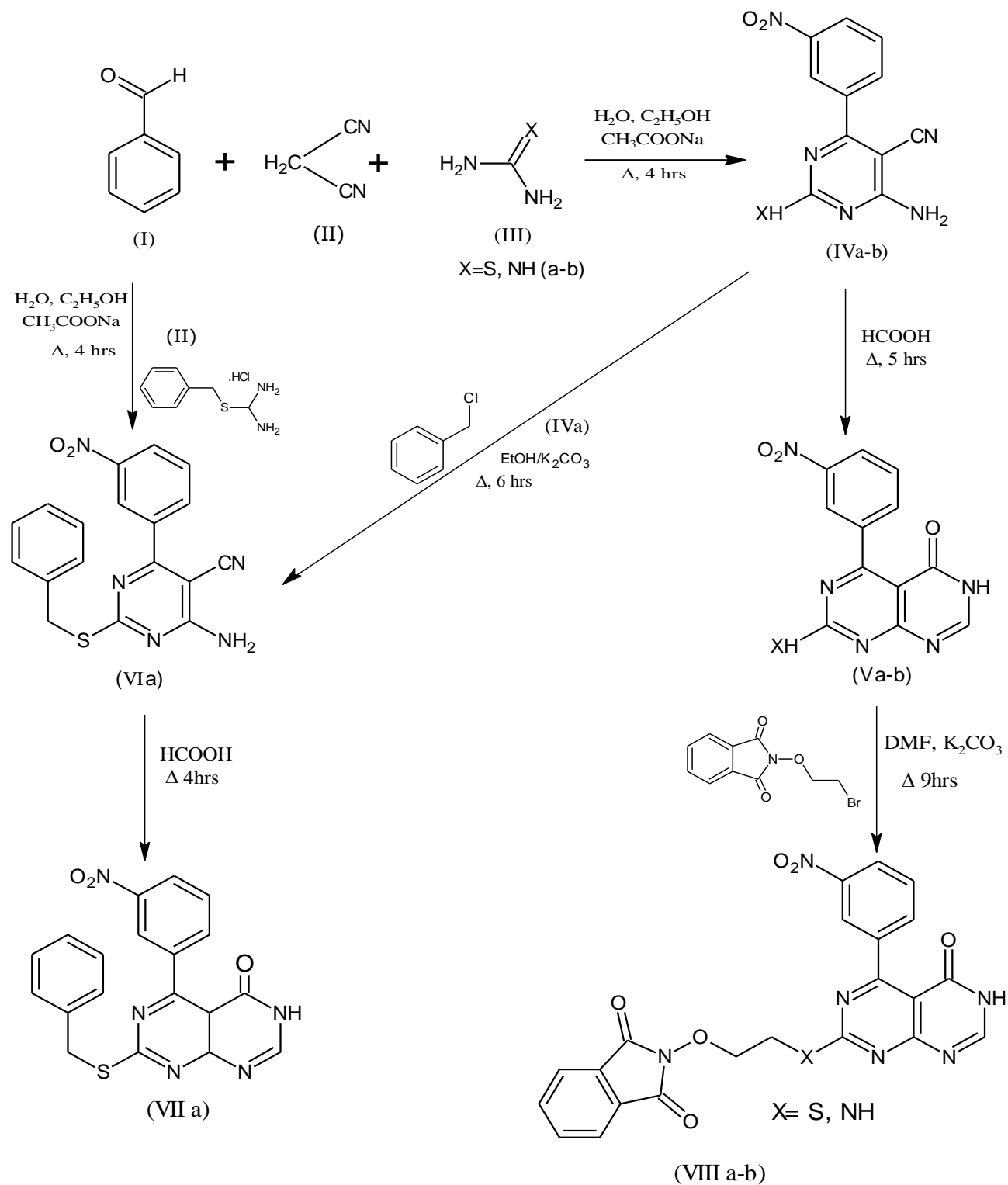
Synthesis of 4-amino-6-(3-nitrophenyl)-2-sulfanylpuridine-5-carbonitrile (IV a): An equimolar mixture of aromatic aldehyde (0.01 mole), malononitrile (0.01 mole), Sodium acetate (0.01 mole) & Thiourea (0.01 mole) was refluxed in EtOH:Water (2:8) (20 ml) for 3 hrs. After completion of reaction (TLC) the reaction mixture was removed from the heating source and kept at room temperature for at least 2 hrs. A crystalline product was obtained which was collected by filtration, washed with chilled EtOH, dried & recrystallization from Ethanol to give desired pure product.

Yield: 65%; M.P.: 260-262°C; IR (KBr) cm^{-1} : 3426, 3393 (NH_2 str.), 3032 (C-H str., Ar-H), 2222 (CN str.), 1650 (C=N str.). ^1H NMR (DMSO) δ : 3.32 (s, 1H, SH), 4.14 (s, 2H, NH_2), 7.61-7.96 (m, 4H, Ar-H). MS: m/z; 273, Anal. calcd for $\text{C}_{11}\text{H}_7\text{N}_5\text{O}_2\text{S}$: C 48.35; H 2.58; N 25.63; O 11.71; S 11.73%. Found: C 48.31; H 2.52; N 25.60; O 11.73; S 11.69%.

Synthesis of 3, 5-diamino-3'-nitrobiphenyl-2-carbonitrile (IV b): An equimolar mixture of aromatic aldehyde (0.01 mole), malononitrile (0.01 mole), Sodium acetate (0.01 mole) & Guanidine nitrate (0.01 mole) was refluxed in EtOH:Water (2:8) (20 ml) for 3 hrs. After completion of reaction (TLC) the reaction mixture was removed from the heating source and kept at room temperature for at least 2 hrs. Yellow coloured crystals were obtained which was collected by filtration, washed with chilled EtOH, dried & recrystallization from Ethanol to give desired pure product.

Yield: 60%; M.P.: 146-148°C IR (KBr) cm^{-1} : 3412, 3397 (NH_2 str.), 3032 (C-H str., Ar-H), 2222 (CN str.), 1643 (C=N str.). ^1H NMR (DMSO) δ : 4.48 (s, 4H, NH_2) 7.63-7.96 (m, 4H, Ar-H). MS: m/z; 256,. Anal. calcd for $\text{C}_{11}\text{H}_8\text{N}_6\text{O}_2$: C 51.56; H 3.15; N 32.80; O 12.49%, Found: C 51.49; H 3.11; N 32.76; O 12.52%.

Synthesis of 7-Substituted-5-(3-nitrophenyl) pyrimido [4, 5-d] pyrimidin-4(3H)-one (V a-b): Compound (IV a-b) (0.01 mole) was taken in 250 ml RB flask and formic acid (40 ml) was added to it. The reaction flask is attached to reflux condenser and the reaction mixture is heated to reflux for 5 hrs. After completion of reaction (TLC) reaction mixture was cooled and poured in crushed ice. A yellow brown colour precipitate obtained which was separated by filtration, washed with water, dried & recrystallized by ethanol to give pure product.



Reaction Scheme

5-(3-nitrophenyl)-7-sulfanylpurimido [4,5-*d*]purimidin-4(3*H*)-one (Va): Yield: 59%; M.P.: 244-248°C IR (KBr) cm⁻¹: 3391 (NH str.), 3200 (SH), 3031 (C-H str., Ar-H), 2936 (C-H Aliphatic) 1692 (CO str.), 1645 (C=N str.). ¹H NMR (DMSO) δ: 1.56 (s, 1H, CH) 3.32 (s, 1H, SH) 8.56 (s, 1H, NH), 7.63-7.96 (m, 4H, Ar-H), Anal. calcd for C₁₂H₇N₅O₃S: C 47.84; H 2.34; N 23.25; O 15.93; S 10.64%. Found: C 47.72; H 2.34; N 23.28; O 15.87; S 10.54%.

7-amino-5-(3-nitrophenyl) purimido [4,5-*d*]purimidin-4(3*H*)-one (Vb): Yield: 62%; M.P.: 220-225°C IR (KBr) cm⁻¹: 3411, 3389 (NH₂ & NH str.), 3032 (C-H str., Ar-H), 2931 (C-H Aliphatic) 1696 (CO str.), 1647 (C=N str.). ¹H NMR (DMSO) δ: 1.48 (s, 1H, CH) 4.56 (s, 2H, NH₂) 8.41 (s, 1H, NH), 7.26-7.91 (m, 4H, Ar-H), Anal. calcd for C₁₂H₈N₆O₃: C 50.71; H 2.84; N 29.57; O 16.89%. Found: C 50.55; H 2.76; N 29.51; O 16.91%.

Synthesis of 4-amino-2-(benzylsulfanyl)-6-(3-nitrophenyl) purimidine-5-carbonitrile (VIa): Method A (from IVa): Compound (IV a) (0.01 mole) was taken in 250 ml RB flask and benzyl chloride (0.01 mole) and K₂CO₃ (0.01 mole) was taken in 20 ml DMF and the reaction mixture was refluxed for 6 hrs. After completion of the reaction (TLC) the reaction mixture was cooled and poured in crushed ice. A precipitated product separated which was collected by filtration, washed with water dried and recrystallized from ethanol.

Method B (Direct method): An equimolar mixture of aromatic aldehyde (0.01 mole), malononitrile (0.01 mole), Sodium acetate (0.01 mole) & phenylisothiouranium chloride (0.01 mole) was refluxed in EtOH:Water (2:8) (20 ml) for 3 hrs. After completion of reaction (TLC) the reaction mixture was removed from the heating source and kept at room temperature for at least 2 hrs. A crystalline product was obtained which was collected by filtration, washed with water, dried & recrystallization from Ethanol to give desired pure product.

Yield: 65%; M.P.: 146-148°C IR (KBr) cm⁻¹: 3426, 3393 (NH₂ str.), 3032 (C-H str., Ar-H), 2950 (C-H str. Aliphatic), 2222 (CN str.), 1650 (C=N str.). ¹H NMR (CDCl₃) δ: 2.42 (s, 2H, CH₂), 4.14 (s, 2H, NH₂), 7.61-7.96 (m, 9H, Ar-H). MS: m/z; 273, Anal. calcd for C₁₈H₁₃N₅O₂S: C 59.49; H 3.61; N 19.27; O 8.81; S 8.82%. Found: C 59.41; H 3.55; N 19.14; O 8.71; S 8.74%.

Synthesis of 7-(benzylsulfanyl)-5-(3-nitrophenyl)-4a, 8a-dihydropurimido [4,5-*d*]purimidin-4(3*H*)-one (VIIa): Compound (VI a) (0.01 mole) was taken in 250 ml RB flask and formic acid (40 ml) was added to it. The reaction flask is attached to reflux condenser and the reaction mixture is heated to reflux for 4 hrs. After completion of reaction (TLC) reaction mixture was cooled and poured in crushed ice. A yellow brown colour precipitate obtained which was separated by filtration, washed with water, dried & recrystallized by ethanol to give pure product.

Yield: 55%; M.P.: 162-164°C IR (KBr) cm⁻¹: 3326 (NH str.), 3072 (C-H str., Ar-H), 2910 (C-H str. Aliphatic), 1692 (C=O Str.), 1650 (C=N str.). ¹H NMR (CDCl₃) δ: 4.12 (s, 2H, CH₂), 1.31 (1H, CH) 9.14 (s, 1H, NH), 7.14-7.96 (m, 9H, Ar-H). MS: m/z; 393, Anal. calcd for C₁₉H₁₅N₅O₃S: C 58.01; H 3.84; N 17.80; O 12.20; S 8.15%. Found: C 58.07; H 3.81; N 17.86; O 12.17; S 8.11%.

Synthesis of 2-(2-[[4-(3-nitrophenyl)-5-oxo-5,6-dihydropurimido[4,5-*d*]purimidin-2-yl] sulfanyl} ethoxy)-1*H*-isoindole-1,3(2*H*)-dione (VIIIa): Bromoethoxy phthallimide (0.01 mole) and compound (Va) (0.01 mole) was dissolved in DMF (20 ml) and Na metal (0.01 mole) is added. Then the reaction

mixture is stirred on a magnetic stirrer for 3 hrs at room temperature. After that the reaction mixture was refluxed for 4 hrs. After completion of reaction (TLC) the reaction mixture is cooled and poured in crushed ice and precipitate of compound (**VIIIa**) was formed which was filtered, washed with cold water, dried and recrystallized from ethanol.

Yield: 65%; M.P.: -224-226°C; IR (KBr) cm⁻¹: 3329 (NH Str.) 3026 (C-H str., Ar-H), 2921 (C-H, str. Aliphatic) 1661 (C=N str.), 1682 (CO-NH str.), 1724, 1694 (CO-N-CO). ¹H NMR (CDCl₃) δ: 1.25 (s, 1H, CH), 4.45 (t, 2H, OCH₂), 3.33 (t, 2H, NCH₂), 7.26-7.81 (m, 18H, Ar-H), 8.91 (s, 1H, NH). MS: m/z; 490. Anal. calcd for C₂₂H₁₄N₆O₆S: C 53.88; H 2.88; N 17.14; O 19.57; S 6.54%. Found: C 53.83; H 2.84; N 17.16; O 19.52; S 6.51%.

Synthesis of 2-(2-[[4-(3-nitrophenyl)-5-oxo-5,6-dihydropyrimido[4,5-d]pyrimidin-2-yl] amino} ethoxy) -1*H*-isoindole-1,3(2*H*)-dione (VIIIb**):** Bromoethoxy phthallimide (0.01 mole) and compound (**Vb**) (0.01 mole) was dissolved in DMF (20 ml) and Na metal (0.01 mole) is added. Then the reaction mixture is stirred on a magnetic stirrer for 3 hrs at room temperature. After that the reaction mixture was refluxed for 4 hrs. After completion of reaction (TLC) the reaction mixture is cooled and poured in crushed ice and precipitate of compound (**VIIIb**) was formed which was filtered, washed with cold water, dried and recrystallized from ethanol.

Yield: 61%; M.P.: -183-186°C; IR (KBr) cm⁻¹: 3310 (NH Str.) 3015 (C-H str., Ar-H), 2931 (C-H, str. Aliphatic) 1645 (C=N str.), 1691 (CO-NH str.), 1730, 1699 (CO-N-CO). ¹H NMR (CDCl₃/DMSO) δ: 4.14 (t, 2H, OCH₂), 2.93 (t, 2H, NCH₂), 7.14-8.12 (m, 18H, Ar-H), 8.91 (s, 1H, NH), 6.11 (s, 1H, NH). MS: m/z; 473. Anal. calcd for C₂₂H₁₅N₇O₆: C 55.82; H 3.19; N 20.71; O 20.28%. Found: C 55.80; H 3.15; N 20.73; O 20.26%.

ANTIMICROBIAL ACTIVITY

All the synthesized compounds of present work were *in vitro* screened for the antibacterial and antifungal activity using 500 ppm concentration in DMF by cup and well method. The micro-organisms *Bacillus subtilis*, *Proteus mirabilis*, *Klebsiella pneumoniae*, *Escherichia coli* were used as antibacterial strains and *Candida albicans* and *Aspergillus fumigatus* were used as fungal strains. The activity is presented as zone of inhibition in mm and compared with activity of controls C1 and C2 (for antibacterial activity C1= ciprofloxacin for antifungal activity C2=flucanazole) to give activity index value (**Table 1**).

All the compounds showed poor activity against *K pneumonia* and *E. coli* whereas moderate to strong activity was shown against *P. mirabilis* and *B. subtilis*. Activity index value against *P. mirabilis* and *B. subtilis* was more than one for majority of compounds. It was interesting to note that all the compounds showed stronger activity than the standard used against *Candida albicans* and *Aspergillus fumigatus*. It was concluded from the activity study that compound **VIIIa** & **VIIIb** was found to be the strongest amongst all synthesized compounds. Compounds under study showed more comprehensive fungus-inhibiting properties than that of the bacterial. Even two folds antifungal activity was observed for these compared to standard.

Table 1: Antimicrobial activity of the synthesized compounds (**IVa-b**), (**Va-b**), (**VIa**), (**VIIa**), (**VIIIa-b**)

Antibacterial activity					Antifungal activity	
S.No.	Bacillus Subtilis	Protius Mirabilis	Klebsilla Pneumonia	Escherichia Coli	Candida Albicans	Aspergillus Fumigatus
IVa	17 (0.98)	18 (1.11)	19 (0.97)	17 (1.94)	18 (0.90)	20 (1.01)
IVb	21(1.18)	20 (0.45)	20 (0.41)	21(1.03)	19 (1.08)	18 (1.02)
Va	20 (0.57)	21 (1.13)	21 (1.08)	20 (1.25)	17 (1.05)	18 (1.09)
Vb	24(1.34)	23 (1.29)	20 (1.37)	21(1.03)	18 (1.08)	19 (1.12)
VIa	16 (1.38)	18 (1.11)	19 (0.97)	17 (0.94)	18 (0.90)	20 (1.11)
VIIa	24(1.34)	23 (1.29)	20 (0.41)	21(1.03)	16 (1.08)	19 (1.02)
VIIIa	24 (1.37)	24 (1.29)	23 (1.18)	24 (1.25)	23 (1.05)	24 (1.13)
VIIIb	24(1.34)	23 (1.29)	20 (1.37)	21(1.03)	22 (1.08)	22 (1.02)
C1	18	17	18	18	-	-
C2	-	-	-	-	20	20

(Activity index) = Inhibition zone of compound/Inhibition zone of the standard drug

For antibacterial activity: C1 = Ciprofloxacin

For antifungal activity: C2 = flucanazole

CONCLUSION

In the synthesized compounds **IVb**, **Va**, **Vb**, **VIIa**, **VIIIa**, **VIIIb** give good activity and others show moderate activity against all four bacterial and **VIIIa** & **VIIIb** give good activity against two fungal.

ACKNOWLEDGEMENT

Authors are thankful to the Head, Department of Chemistry, Mewar University, Gangrar, Chittorgarh for providing Laboratory facilities and to the Director, MRC, MNIT, Jaipur, India for providing spectral and analytical data.

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On line publication Date: 26.12.2016